

0040-4039(93)E0392-W

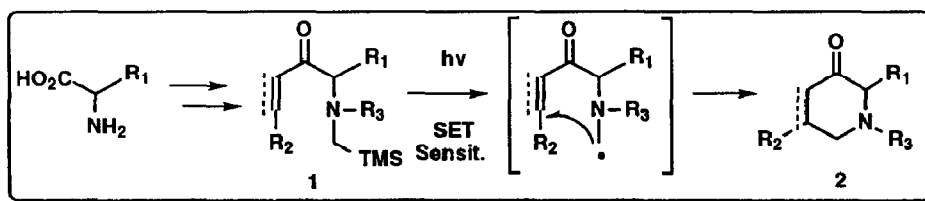
A Novel Method for Synthesis of Functionalized Piperidines

Seock Kyu Kim and Patrick S. Mariano*
 Department of Chemistry and Biochemistry
 University of Maryland, College Park, MD 20742

Abstract. A new methodology for synthesis of functionalized piperidines which employs photoinduced radical cyclization of α -silylamino-enones and ynones is described.

Earlier, we described the results of studies which showed that α -silylamino α,β -unsaturated ketone systems upon direct irradiation undergo SET-initiated photoaddition and photocyclization reactions.¹ In addition, we uncovered an SET-photosensitization methodology to promote related α -amino radical cyclization processes. Several of the advantages of the SET-sensitization vs. direct irradiation protocol were highlighted in this early work¹ and in more recent studies of silylamino- and silylamido-2,5-cyclohexadienone photocyclization reactions.² A program designed to explore the synthetic potential of this chemistry has led to a recent investigation of a new strategy for synthesis of functionalized piperidines (Scheme 1). We report here the preliminary results of this effort which show that α -silylamino-enones and -ynones **1** can be prepared from an amino acid precursor and that 6-endo radical cyclizations of these substances occur under SET-photosensitization conditions to produce piperidines **2** with high degrees of regiochemical and stereochemical control.

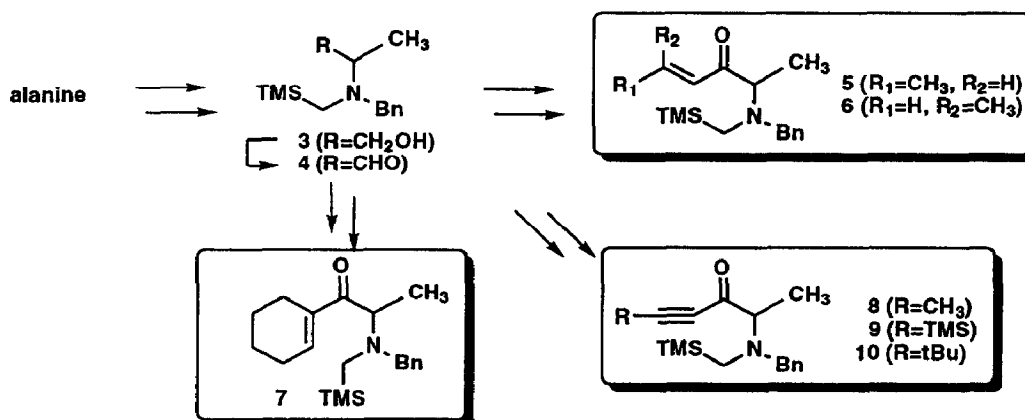
Scheme 1.



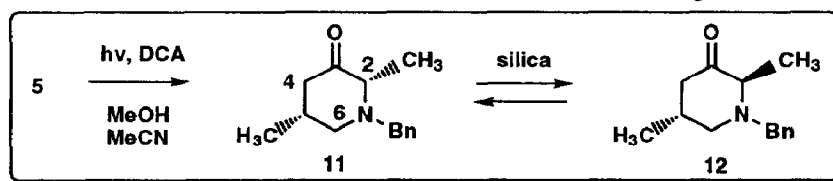
Silylamino-enones **5-7** and -ynones **8-10** were used to test various features of this piperidine synthesis strategy. Preparation of these substances started with sequential conversion of L-alanine to amino-alcohol **3** and -aldehyde **4**^{3a} (Scheme 2). Addition of $\text{LiC}\equiv\text{CCH}_3$ to **4** followed either by Red-Al reduction and Swern oxidation gives **5**^{3b} (28%) or by hydrogenation (Lindlar) and TPAP/NMO oxidation⁴ gives **6**^{3c} (10%). In a similar fashion, addition of 1-cyclohexenyl lithium to **4** followed by Swern oxidation gives **7**^{3d} (32%). Finally, independent addition of $\text{LiC}\equiv\text{CCH}_3$, $\text{LiC}\equiv\text{CTMS}$ and $\text{LiC}\equiv\text{CtBu}$ to **4** followed by Swern oxidation of each of the resulting ynols provides the respective ynones **8-10**. By these procedures the amino-enones are prepared as

chromatographically pure substances. The ynones ($> ca.$ 80% purity), on the other hand, are labile materials and, therefore, must be used in the photoreactions described below without purification.

Scheme 2.

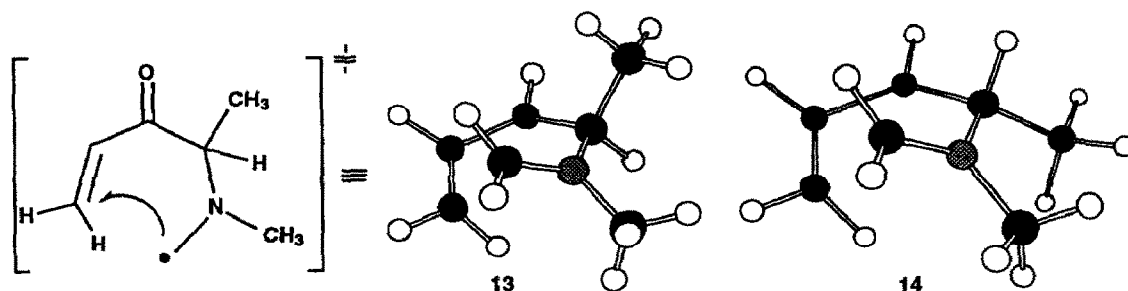


SET-photosensitized reactions were conducted by irradiation ($\lambda > 320$ nm) of 9,10-dicyanoanthracene (DCA) (*ca.* 0.1 mM) in 15% MeOH-MeCN solutions of the enones or ynones (*ca.* 1 mM). Under these conditions, the *trans*-propenyl amino-ketone **5** is transformed to a single, diastereomerically pure piperidine **11** in a yield which varies with the extent of conversion (*e.g.* 73% at 6% conversion of **5**, 23% at 73% conversion). Chromatography of the photolysate on silica gel yields an equilibrium mixture of **11** and its stereoisomer **12** in an *ca.* 1:1 ratio. The piperidines, **11** and **12**, are characterized as the respective *cis*- and *trans*-stereoisomers on the basis of their distinctive spectroscopic properties. For example, the ^{13}C NMR chemical shifts of C-6 and C-4 in **11** (52.7 and 44.7 ppm) and **12** (57.6 and 47.0 ppm) are characteristic of gauche-effects associated with the axial C-2 methyl conformer which contributes to the *cis*-isomer **11**. Moreover, the silica gel induced conversion of **11** to **12** by enolization suggests their C-2 epimeric relationship. In this process as well as those discussed below, both glc and ^1H NMR analyses of the crude photolysates at varying conversion indicate that $< 95\%$ of the alternative stereo- or regio-isomers are produced.

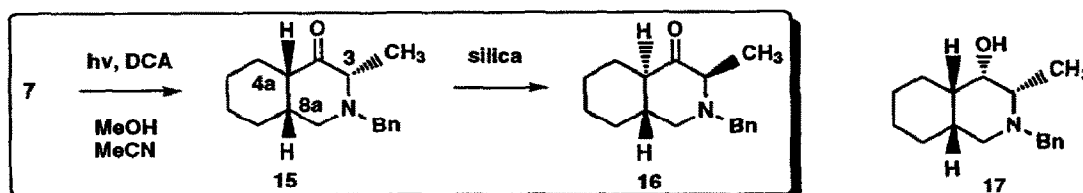


The high degree of *cis*-diastereoselectivity for the radical cyclization step in the conversion of **5** to **11** is consistent with a predictive model. Specifically, the model transition states, **13** and **14** were derived by molecular mechanics minimization (Macromodel/MM2) using constrained approach distances and angles calculated in earlier⁵ *ab initio* treatments of radical addition processes. The greatly different energies of **13** (17.5 kcal/mol) vs. **14** (19.5 kcal/mol), resulting from a combination of A^{1,2}-strain (N-CH₃-C-CH₃) and transannular steric (H-H vs. H-CH₃) interactions, indicate that radical cyclization in the photoreaction of **5** should lead to preferential formation of the

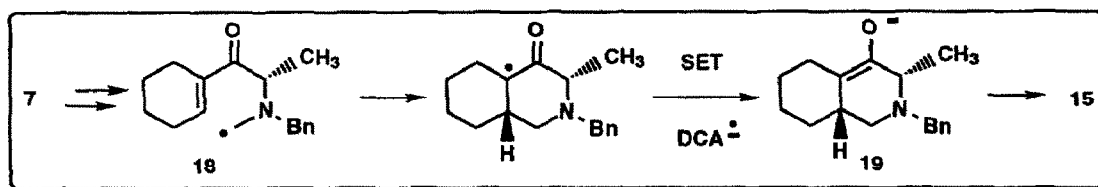
α -keto radical precursor of **11**. Moreover, similar considerations suggest that the *cis*-propenyl amino ketone **6** should resist radical cyclization owing to a highly congested transition state. This seems to be a correct forecast since no piperidine products are formed when **6** is subjected to the DCA-sensitized irradiation conditions.



The high diastereoselectivity of this 6-endo radical cyclization is further demonstrated by the SET-sensitized photochemistry of the cyclohexenyl ketone **7**. DCA-induced photoreaction of this substance leads to efficient (69% at 86% conversion) and stereoselective formation of a single decahydroisoquinoline **15**. The *cis,cis*-stereochemistry assignment to this substance follows from its characteristic spectroscopic properties (eg., ^{13}C NMR 47.9 (C-4a), 37.6 (C-8a); ^1H NMR 2.73 (dd, $J=11.7, 3.7$ Hz, H-1ax), 2.51 (dd, $J=11.7, 2.6$ Hz, H-1eq)), its facile epimerization on silica gel to form the *trans*-fused hydroisoquinoline **16** (eg., ^{13}C NMR 49.3 (C-4a), 42.1 (C-8a); ^1H NMR 2.67 (dd, $J=12.3, 10.8$ Hz, H-1ax), 2.54 (dd, $J=12.3, 4.1$ Hz, H-1eq)) and its stereoselective reduction to give a single alcohol stereoisomer **17**.

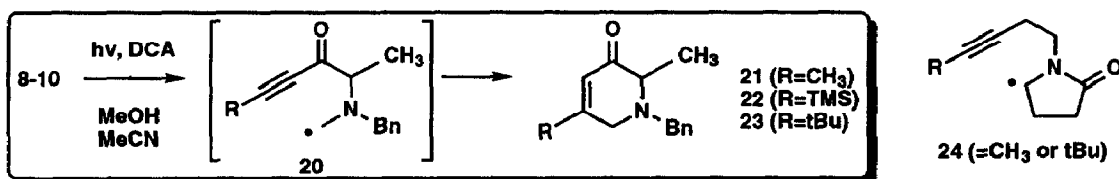


The *cis*-stereochemistry at C-3 and C-8a in **15** is consistent with the results of transition state modeling for cyclization of the radical **18** and the *cis*-relationship at C-4a and C-8a is as expected based on least hindered approach control in protonation of the final intermediate, enolate **19**.



Finally, the *endo*-radical cyclization protocol is applicable to the synthesis of unsaturated piperidinones. Accordingly, DCA-sensitized irradiations of the amino-ynones **8-10** in MeOH-MeCN

leads to production of the respective piperidines **21-23** (ca. 25% yields starting with ynol precursors of **8-10**). 6-Endo rather than 5-exo radical cyclization regiochemistry is adhered to in all three cases. This result stands in contrast to related cyclizations of the α -amido alkynyl radicals **24** probed earlier by Hart⁶ where the 6-endo route is followed preferentially by the propynyl system but where the t-butyl-ethynyl analog reacts by a 5-exo mode exclusively. One of several possible reasons for this difference might be that α -amino-ynone (**20**) as compared to α -amido-alkynyl (**24**) radical cyclization reactions might be influenced to a greater extent by polar (*i.e.*, FMO-coefficients) rather than steric effects.



The results summarized above show the viability of SET-photoinduced, 6-endo radical cyclization processes in methodology for functionalized piperidine synthesis. Further efforts are needed to improve the efficiency and develop the potential enantiospecificity of the strategy.

Acknowledgment. This study was supported by a grant from the NIH (GM-27251).

References

- (1) (a) Hasegawa, E.; Xu, W.; Mariano, P.S.; Yoon, U.C.; Kim, J.U. *J. Am. Chem. Soc.*, **1988**, *110*, 8099; (b) Jeon, Y.T.; Lee, C.P.; Mariano, P.S. *ibid.*, **1991**, *113*, 8847; (c) Xu, W.; Zhang, X.M.; Mariano, P.S. *ibid.*, **1991**, *113*, 8863; (d) Yoon, U.C.; Mariano, P.S. *Acc. Chem. Res.*, **1992**, *25*, 233.
- (2) Jung, Y.S.; Swartz, W.H.; Mariano, P.S.; Green, N.J.; Schultz, A.G. *J. Org. Chem.*, **1992**, *57*, 6037.
- (3) (a) Aldehyde **4** ($[\alpha]_D^{29} +1.5^\circ$) derived from L-alanine has a >95% ee as determined by Mosher ester analysis of the alcohol **3** obtained by NaBH₄ reduction of **4**; (b) $[\alpha]_D^{25} -6.5^\circ$, 22% ee determined by reduction (NaBH₄, CeCl₃) of **5** which gives the anti and syn alcohol diastereomers in a 1:2 ratio. Mosher ester analysis of the minor isomer ($[\alpha]_D^{25} +6.4^\circ$) gives a 21% ee. This same anti alcohol isomer ($[\alpha]_D^{25} +30.9^\circ$, ca. 90% ee) is formed as the near exclusive product of sequential reaction of aldehyde **4** with LiC \equiv CCH₃ and Red-Al; (c) $[\alpha]_D^{29} -0.6^\circ$; (d) $[\alpha]_D^{28} -25.7^\circ$.
- (4) Griffith, W.P.; Ley, S.V.; Whitcombe, G.P.; White, A.D. *J. Chem. Soc. Chem. Commun.*, **1987**, 1625.
- (5) Damm, W.; Giese, B.; Hartung, J.; Hasskerl, T.; Houk, K.N.; Huter, O.; Zipse, H. *J. Am. Chem. Soc.*, **1992**, *114*, 4067. The models use planar α -amino radical centers and approach distances of 2.3 Å and angles of 102° and other constraints recommended by these authors.
- (6) Choi, J.K.; Hart, D.J. *Tetrahedron*, **1985**, *41*, 3959.

(Received in USA 8 September 1993; revised 19 November 1993; accepted 10 December 1993)